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ANTIMONYPENTAFLUOROORTHOTELLURATES: Sb(OTeF<sub>5</sub>)<sub>3</sub>, Sb(OTeF<sub>5</sub>)<sub>5</sub> AND SALTS OF Sb(OTeF<sub>5</sub>)<sub>6</sub>

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Antimony(III)pentafluoroorthotellurate has been synthesized from SbF $_3$  and B(OTeF $_5$ ) $_3$ . Contrary to a previous report it is a low melting, sublimable solid (mp = 28°, bp (0.1 torr) = 68°,  $^{19}\mathrm{F}$  - NMR : AB $_4$  spinsystem  $\delta$  (A) = -42.7,  $\delta$  (B) = -38.1, J (AB) = 186 Hz). It reacts with F $_2$ , Cl $_2$  and Br $_2$  to give SbF $_2$ (OTeF $_5$ ) $_3$ , SbCl $_4$ Tsb(OTeF $_5$ ) $_6$  and SbBr $_4$ Tsb(OTeF $_5$ ) $_6$  respectively. Interaction of Xe(OTeF $_5$ ) $_2$  and Sb(OTeF $_5$ ) $_3$  yields Sb(OTeF $_5$ ) $_6$ , which is unstable at room temperature. Salts containing the new anion Sb(OTeF $_5$ ) $_6$  have been synthesized either from Sb(OTeF $_5$ ) $_5$  and a corresponding pentafluoroorthotellurate e.g. Sb(OTeF $_5$ ) $_5$  + NMe $_4$  OTeF $_5$  = NMe $_4$  Sb(OTeF $_5$ ) $_6$  , or from SbCl $_4$  Sb(OTeF $_5$ ) $_6$  and an appropriate chloride SbCl $_4$  + Sb(OTeF $_5$ ) $_6$  - NOCl = SbCl $_5$  + NO+ Sb(OTeF $_5$ ) $_6$  -, or oxidatively, using a mixture of Xe(OTeF $_5$ ) $_2$  and Sb(OTeF $_5$ ) $_5$ , e.g. C $_6$ F $_6$  + 1/2 Xe(OTeF $_5$ ) $_2$  + Sb(OTeF $_5$ ) $_5$  = C $_6$ F $_6$  + Sb(OTeF $_5$ ) $_6$  - + 1/2 Xe.

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PREPARATION AND STEREOCHEMISTRY OF ALKYL TELLURIUM(IV) FLUORIDES AND PENTAFLUOROTELLURATES(VI)

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Alkyltellurium(IV)iodides interact with AgF with the formation of alkyltellurium(IV)fluorides and with AgOTeF, with the formation of the respective pentafluorotellurates(VI). Depending on the number of alkyl groups these compounds show a drastic variation with respect to molecular structure and steric rigidity as revealed by  $^{19}\text{F-}$  and  $^{125}\text{Te-n.m.r.}$  spectroscopy.